Cell Design

Undivided cell Preferred Simple setup



working counter strictions electrods

Quasi-divided cell
Electrode sizes
differ strongly in
undivided cell

Divided cell

Separation of anolyte and catholyte suppress side reactions at counter electrode



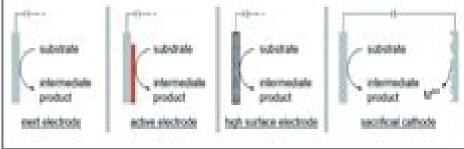
Electrodes

Inert – (Electro-)chemically stable (Pt, graphite, glassy carbon, BDD)

Active – Electrogeneration of insoluble high valent species (Mo, Ni, Pb)

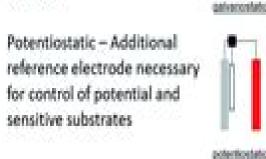
Sacrificial cathode – Consumed during electrolysis (Zn, Al, Mg)

High surface – Area structured (meshes, foams – RVC, graphite felt, Ni foam)



Modes of Operation

Galvanostatic – Two electrode setup with simple DC devices for fast and robust reactions; scalable



Electron Transfer

Direct electrolysis (in-cell) – Heterogenous electron transfer at surface Mediated electrolysis (in-cell) – Electron transfer with soluble reagent for homogenous electron transfer to substrate

Electrochemical catalysis – sub-stoichiometric amounts of charge in ECE^b mechanism

Ex-cell – Electrogeneration of reactive intermediates and addition of substrates after electrolysis



Electro Organic Synthesis Guide

Sabine Zange

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Tutorial Lectures in Electrochemical Engineering and Technology R. C. Alkire,1981 **Bibliography of Electro-organic Syntheses, 1801-1975** R. C. Alkire,1980 <u>Electro-organic Synthesis Technology</u> Mike Krumpelt,1979

Electroorganic Chemistry as a New Tool in Organic Synthesis Tatsuya Shono, 2012-12-06 Although the first electroorganic reaction used in organic synthesis is probably the famous Kolbe electrolysis published in 1849 no other remarkable reactions have been found until the reductive dimerization of acrylonitrile to adipo nitrile was developed by Dr M M Baizer of Monsanto Co in 1964 Since then the electro organic chemistry has been studied extensively with the expectation that it is a new useful tool for finding novel reactions in organic synthesis The purpose of this book is not to give a comprehensive survey of studies on electrochemical reactions of organic compounds but to show that the electro organic chemistry is indeed useful in organic synthesis Thus this book has been written under the following policies 1 Since this monograph is mainly concerned with organic synthesis only few studies carried out from the view point of electrochemical theoretical or analytical chemistry are mentioned 2 Since electroorganic chemistry covers a great variety of reactions the types of reactions described in this book are selected mainly with regard to their application in organic synthesis Simple transformations of functional groups are only described in particular cases and also some well established processes such as the Kolbe electrolysis pinacolic coupling and hydrodimerization are only briefly mentioned 3 Since many reports have already been published for each type of these reactions only a limited number of the relevant papers are cited in this book

Classified Guide to Technical and Commercial Books ,1904 **Technique of Electroorganic Synthesis** Norman L. Weinberg,1974 Peterson's Guide to Graduate Programs in the Physical Sciences and Mathematics ,1991 **Nuclear Science Abstracts** ,1974 Platers' Guide ,1915 Chemical Reaction and Reactor Engineering Carberry,2020-08-26

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